

Experimental Aspects of Neutron Diffraction

William Ratcliff

NCNR





The Nobel Prize in Physics 1994



Clifford G. Shull, MIT, Cambridge, Massachusetts, USA, receives one half of the 1994 Nobel Prize in Physics for development of the neutron diffraction technique.



Shull made use of elastic scattering i.e. of neutrons which change direction without losing energy when they collide with atoms.

Because of the wave nature of neutrons, a diffraction pattern can be recorded which indicates where in the sample the atoms are situated. Even the placing of light elements such as hydrogen in metallic hydrides, or hydrogen, carbon and oxygen in organic substances can be determined.

The pattern also shows how atomic dipoles are oriented in magnetic dipoles, since neutrons are affected by magnetic forces. Shull also made use of this phenomenon in his neutron diffraction technique.



An early (1950) neutron diffractometer with fixed wavelength control here used by E.O. Wollan and C.G. Shull (standing) at Oak Ridge National Laboratory.

Neutrons see more than X-rays

X-rays are scattered by electrons, neutrons by atomic nuclei. With X-rays, for example, it is easier to see atoms that have many electrons. Hydrogen, for example, which has only one electron, is not so easy to see. With neutrons, all kinds of atoms are visible.



Let a neutron diffraction map (showing the positions of the nuclei) over an X-ray diffraction map (giving the distribution of the electrons). You will see that the density is shifted in relation to the position of the nuclei. Since a chemical bond involves a shift in electron position, a direct picture of the chemical bond is obtained in this way.

Neutrons reveal inner stresses

A hole has been punched in an important metal aircraft part. Does the part match up? Neutron diffraction can show how much the distortion is. When a neutron has passed through a material, its charge has changed and hence the internal forces remaining round the hole after it has been punched.

The curves show local expansion forces (positive) and compression forces (negative) in different directions (red, green and blue) in an aircraft part. (See also Gripen.)

Neutrons show what atoms remember

The time interval $t = 0$ shows the positions of the liquid bromine. The other curves show how the positions of the atoms change with time ($1 \text{ ps} = 1 \text{ millionth of a millionth of a second}$) seen with inelastic neutron scattering. Corresponding "memory functions" can also be measured in magnets e.g. near the Curie temperature, the temperature at which magnetic order shifts to disorder.

How it started

Brockhouse and Shull made their pioneering contributions at the first nuclear reactors in the USA and Canada back in the 1940s and 1950s. It was then that the resources of the reactors became available for peacetime research.

... how it continues

Thousands of researchers are now working at the many neutron research centers throughout the world. New and very advanced neutron scattering installations have been built and more are planned in Europe, the USA and Asia. At these super-installations the researchers are studying the structure of new ceramic superconductors, molecular movements on surfaces of interest for catalysts, exhaust cleaning, virus structures and the connection between the structure and the elastic properties of polymers.

Further reading:

- D.J. Hughes: *The Nuclear Reactor as a Research Instrument*, SCIENTIFIC AMERICAN, VOL. 230, AUGUST 1974, p. 23.
- H. Leugler and J.L. Fourey: *The European Spallation Source*, EUROPHYSICS NEWS, VOL. 25, p. 37, 1994.
- Information about the Nobel Prize in Physics 1994 (pressnote), THE ROYAL SWEDISH ACADEMY OF SCIENCES.

Neutrons behave as particles and as waves

The Royal Swedish Academy of Sciences has awarded the 1994 Nobel Prize in Physics for pioneering contributions to the development of neutron scattering techniques for studies of condensed matter.

Bertram N. Brockhouse, McMaster University, Hamilton, Ontario, Canada, receives one half of the 1994 Nobel Prize in Physics for the development of neutron spectroscopy.



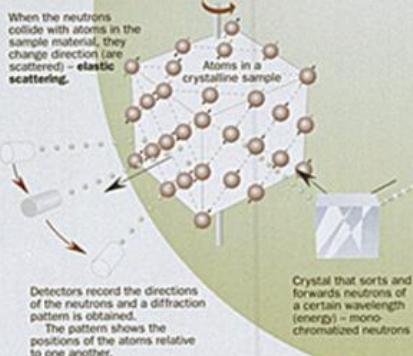
B

Brockhouse made use of inelastic scattering i.e. of neutrons, which change both direction and energy when they collide with atoms. They then start or cancel atomic oscillations in crystals and record movements in liquids and melts. Neutrons can also interact with spin waves in magnets.

With his 3-axis spectrometer Brockhouse measured energies of phonons (atomic vibrations) and magnons (magnetic waves). He also studied how atomic structures in liquids change with time.

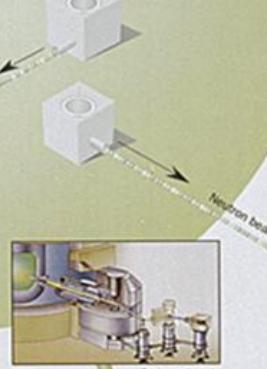
Neutrons reveal structure and dynamics

Neutrons show where atoms are

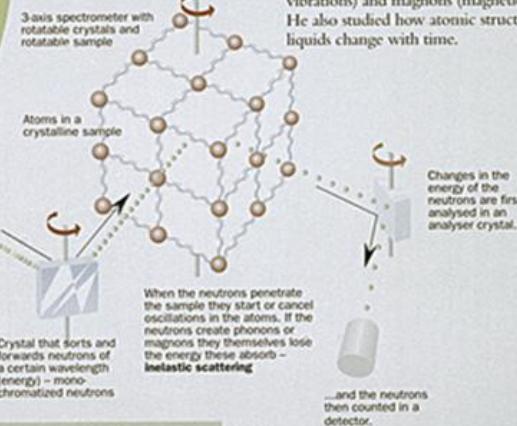


Neutrons bounce against atomic nuclei. They also react to the magnetism of the atoms.

Research reactor



Neutrons show what atoms do



When the neutrons penetrate the sample they start or cancel oscillations in the atoms. If the neutrons create phonons or magnons they themselves lose the energy these absorb – inelastic scattering

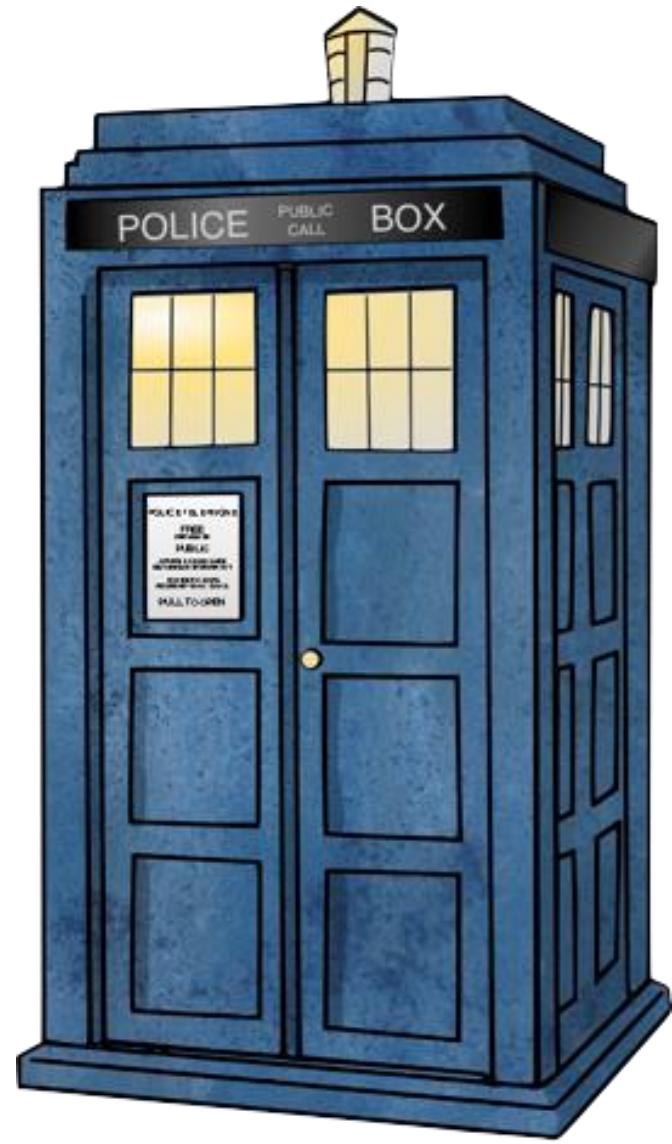
...and the neutrons then counted in a detector.



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AKADEMIEN
THE ROYAL SWEDISH ACADEMY OF SCIENCES

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Fax +46 8 521 95 70, Editorial office: Royal Swedish Academy of Sciences, Stockholm, Mäster Samuelsgatan 10, SE-106 91 Stockholm, Sweden
Royal Swedish Academy of Sciences, Authors: Professor Eric B. Klemm and Professor Carl Nordling, Department of Physics, Uppsala University, Members of the Nobel Committee for Physics, Lund and Illustrations, Åke Lindström, Expressen AB, Printed by Tryckhuset, 1994.





1932 Neutrons Discovered by Chadwick



MAY 15, 1939

PHYSICAL REVIEW

VOLUME 55

On the Magnetic Scattering of Neutrons

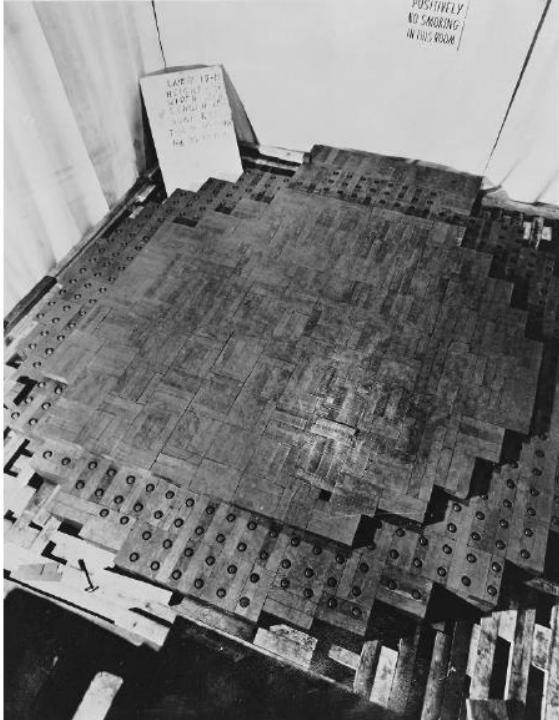
O. HALPERN AND M. H. JOHNSON

New York University, University Heights, New York, New York

(Received December 3, 1938)



1942 Fermi's 30th pile goes critical

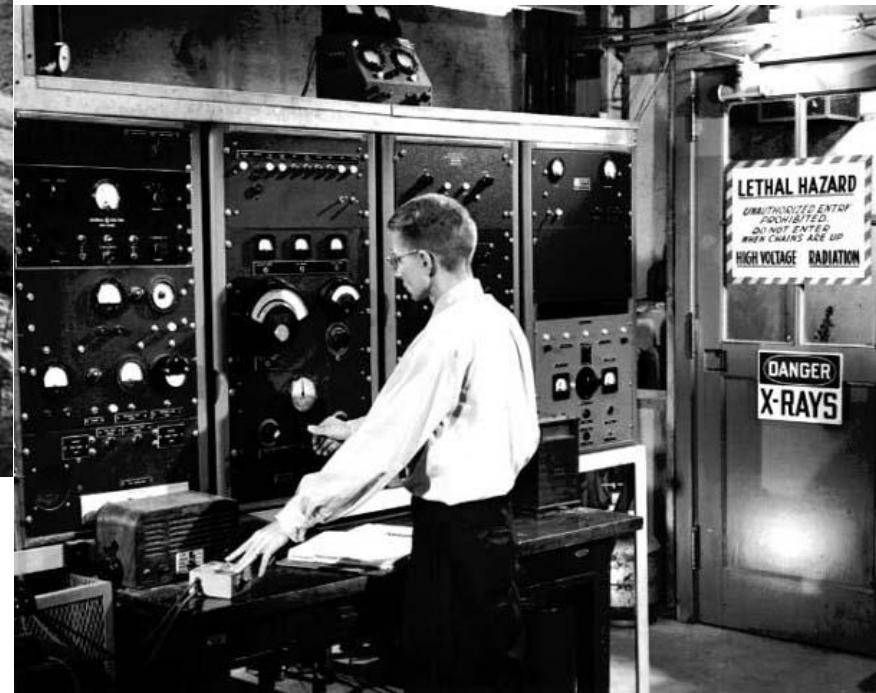
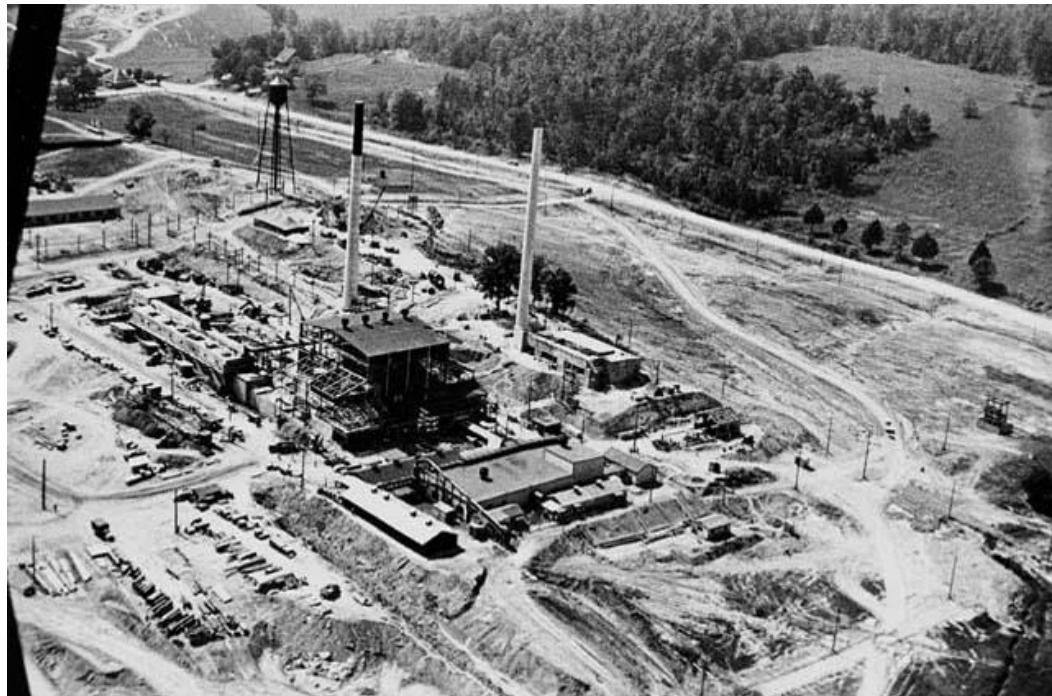


An Account of Oak Ridge National Laboratory's Thirteen Nuclear Reactors August 2009
<https://info.ornl.gov/sites/publications/Files/Pub20808.pdf>



ORNL Graphite Reactor

1943-1963



An Account of Oak Ridge National Laboratory's Thirteen Nuclear Reactors August 2009
<https://info.ornl.gov/sites/publications/Files/Pub20808.pdf>

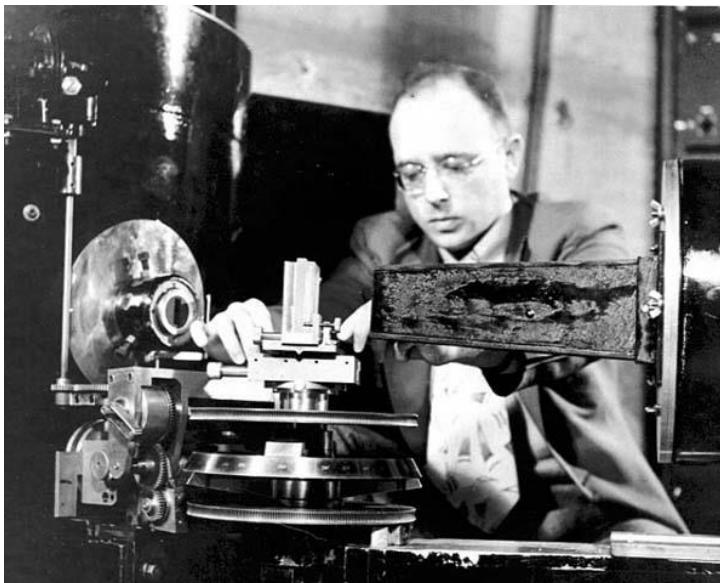


The Diffraction of Neutrons by Crystalline Powders

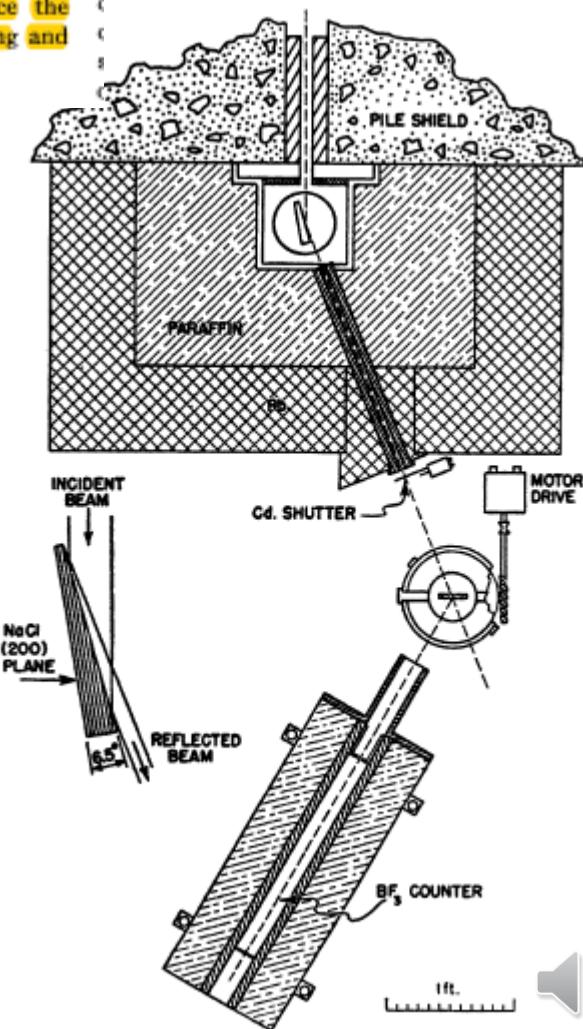
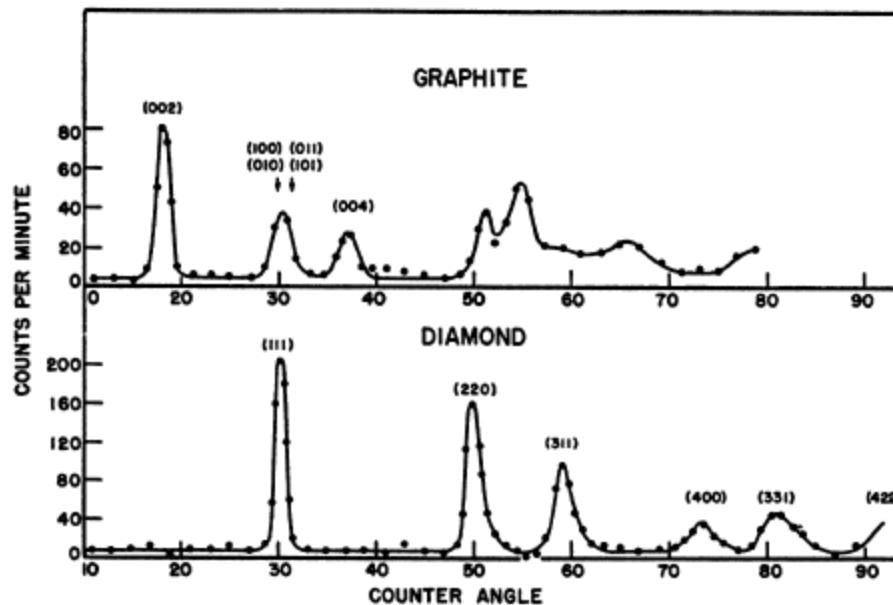
E. O. WOLLAN AND C. G. SHULL

Oak Ridge National Laboratory, Oak Ridge, Tennessee

(Received January 5, 1948)



angular range. The advantage of automatic operation cannot be overestimated since the best data can be taken during the evening and night shifts.

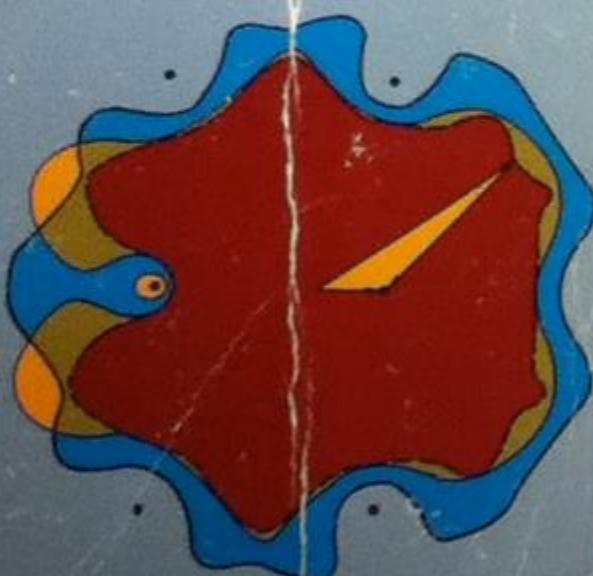




how**stuff**works
It's good to know



INTRODUCTION TO
THE THEORY OF
THERMAL
NEUTRON
SCATTERING



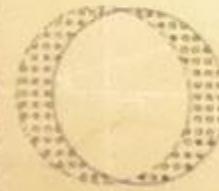
G.L. Squires

Harald Ibach Hans Lüth

Solid-State Physics

An Introduction to Principles of Materials Science

Second Edition



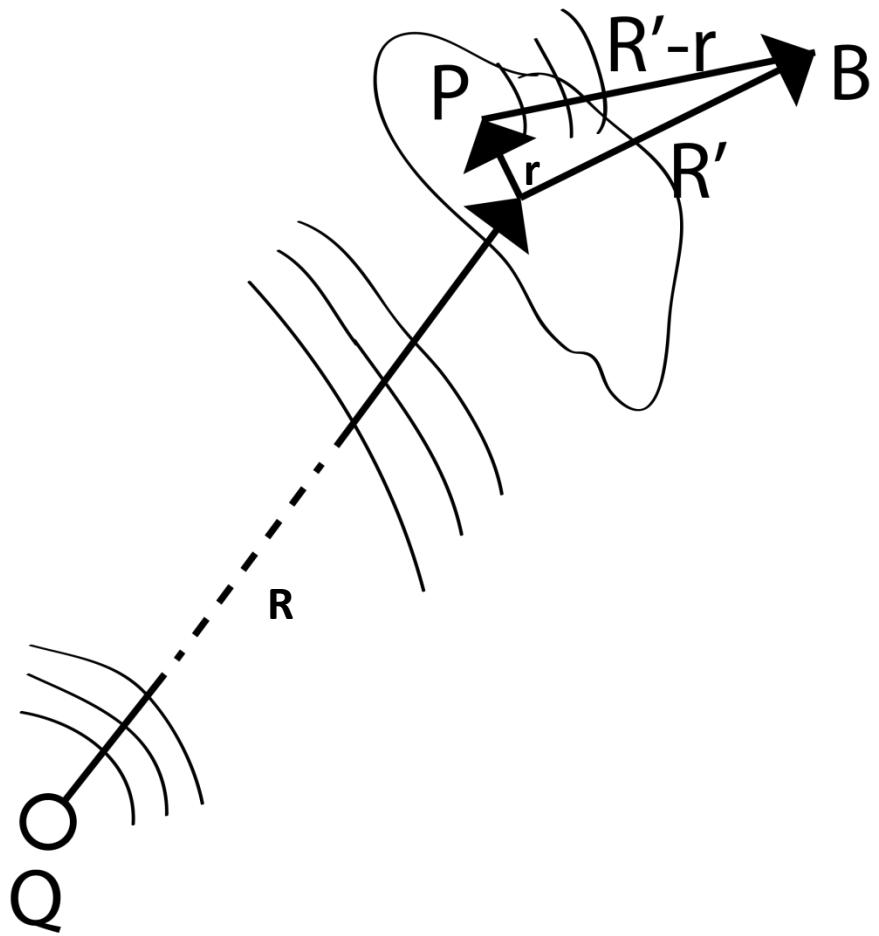
Springer



Crystallography and the reciprocal space

<http://toutestquantique.fr/en/>





$$A_P = A_0 e^{i \vec{k}_0 \bullet (\vec{R} + \vec{r}) - i \omega_0 t}$$

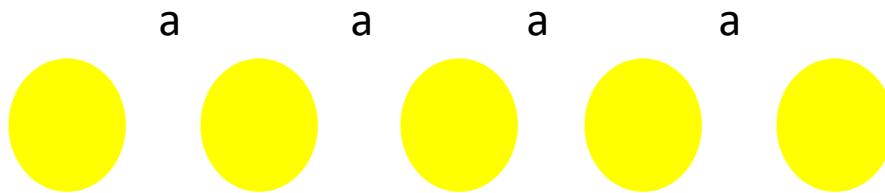
$$A_B = A_P(r, t) \rho(r) \frac{e^{i \vec{k} \bullet (\vec{R}' - \vec{r})}}{|R' - r|}$$

$$A_B = A_P(R' \gg r, t) \rho(r) \frac{e^{i \vec{k} \bullet (\vec{R}' - \vec{r})}}{|R'|}$$

$$I(K) \propto |A_B|^2 \propto \left| \int \rho(r) e^{i \vec{K} \bullet \vec{r}} dr \right|^2$$



Reciprocal Space



$$\rho(x) = \rho(x + na)$$

$$\rho(x) = \sum_n \rho_n e^{i(n2\pi/a)x}$$

$$\rho(\vec{r}) = \sum_n \rho_{\vec{G}} e^{i\vec{G} \bullet \vec{r}} \quad \vec{r}_n = n_1 \vec{a}_1 + n_2 \vec{a}_2 + n_3 \vec{a}_3$$

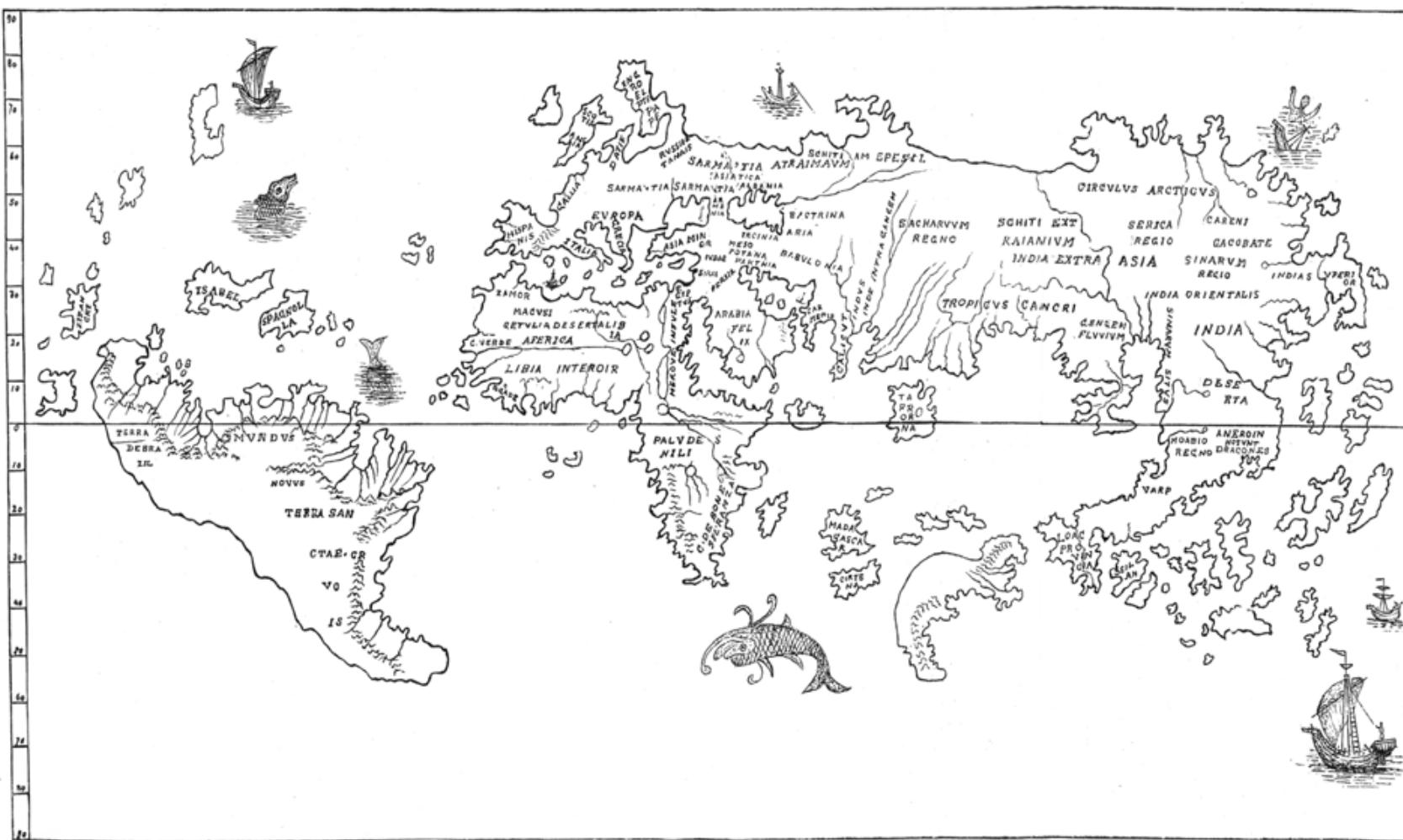
$$\vec{G} \bullet \vec{r} = 2\pi m$$

$$g_1 = 2\pi \frac{\vec{a}_2 \times \vec{a}_3}{\vec{a}_1 \bullet (\vec{a}_2 \times \vec{a}_3)}$$



Here there be dragons...

THE LENOX GLOBE



The Hunt-Lenox Globe, as transcribed by B.F. da Costa

<https://www.theatlantic.com/technology/archive/2013/12/no-old-maps-actually-say-here-be-dragons/282267/>



$$I(K) \propto |A_B|^2 \propto \left| \int \rho(r) e^{-i\vec{K} \bullet \vec{r}} dr \right|^2$$



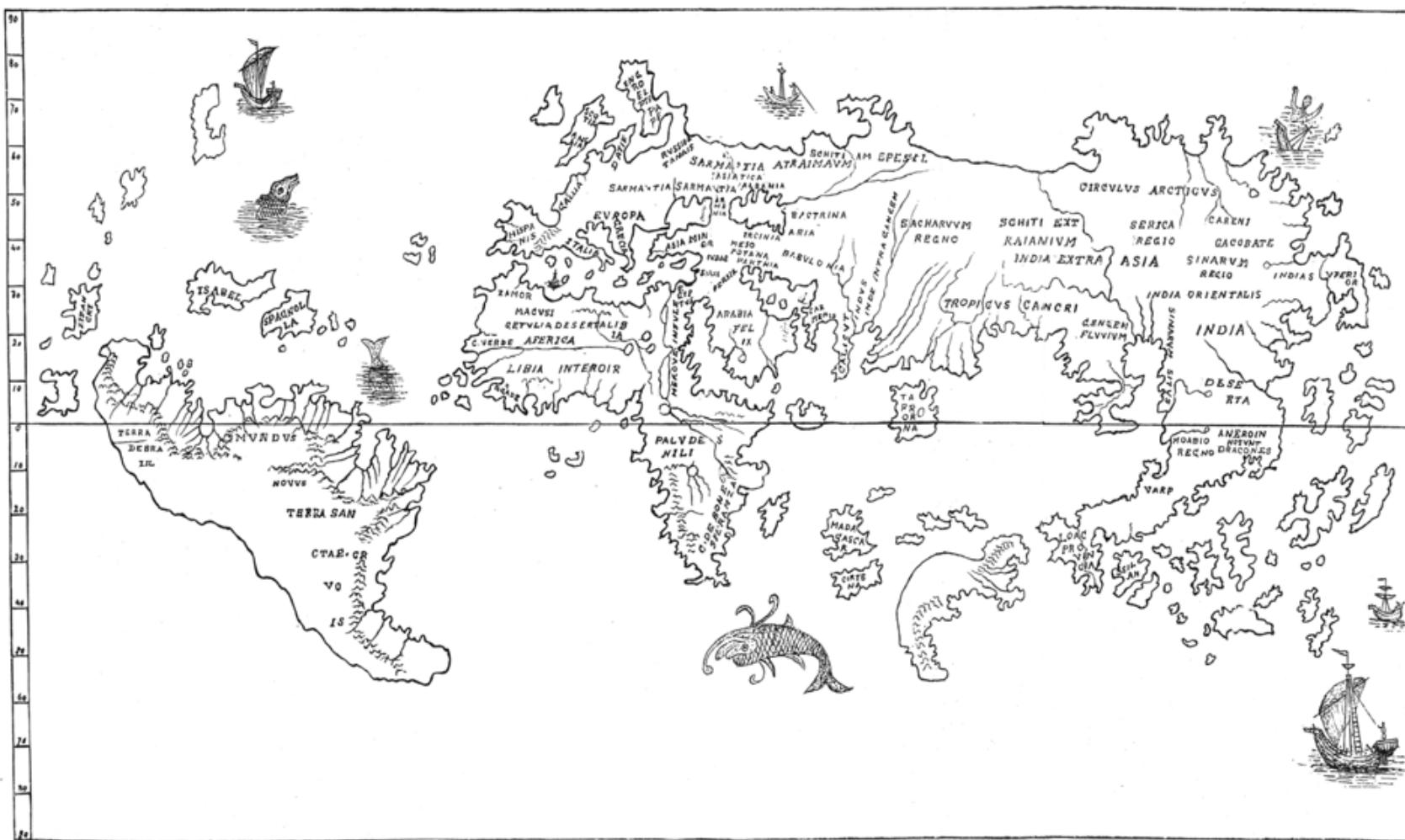
$$I(K) \propto \left| \sum_{\vec{G}} \rho_{\vec{G}}(r) \int e^{i(\vec{G}-\vec{K}) \bullet \vec{r}} dr \right|^2$$

$$\int e^{i(\vec{G}-\vec{K}) \bullet \vec{r}} dr = \begin{cases} V & \text{for } \vec{G} = \vec{K} \\ \sim 0 & \text{otherwise} \end{cases} \quad \text{Laue Condition}$$



Here there be dragons...

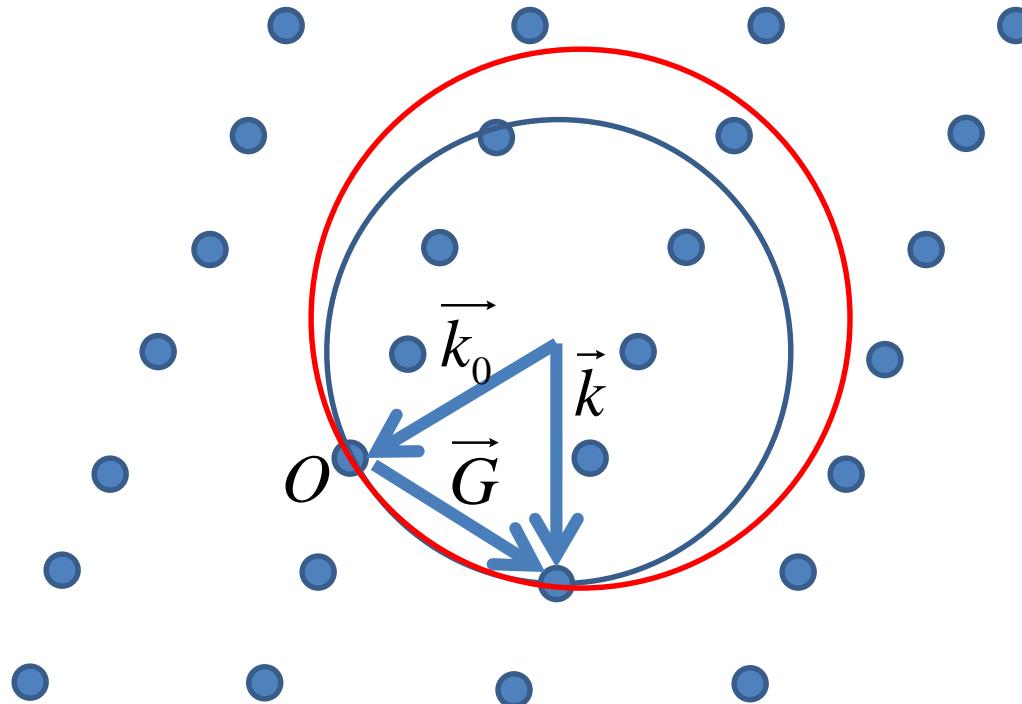
THE LENOX GLOBE



The Hunt-Lenox Globe, as transcribed by B.F. da Costa



Ewald Sphere

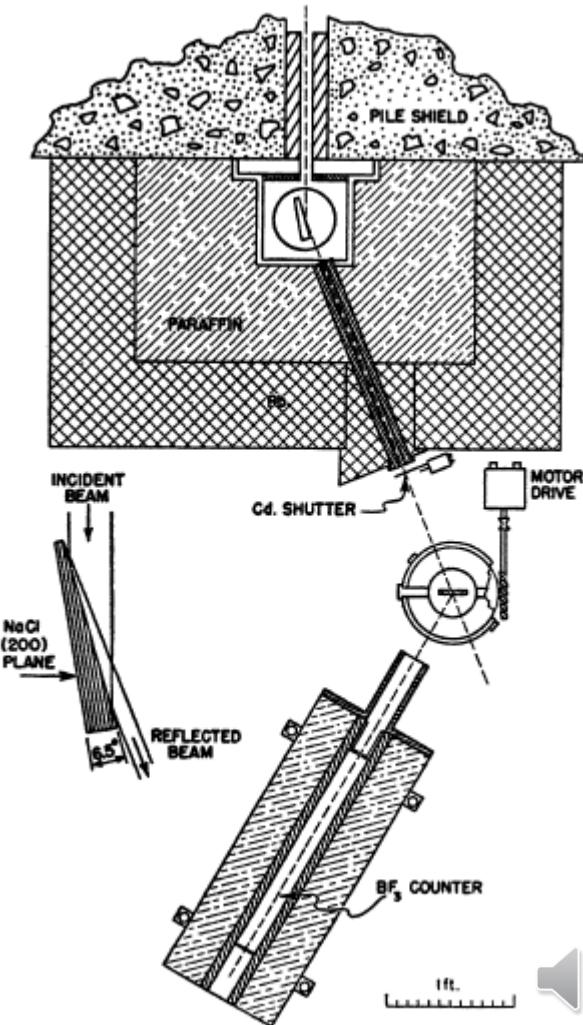
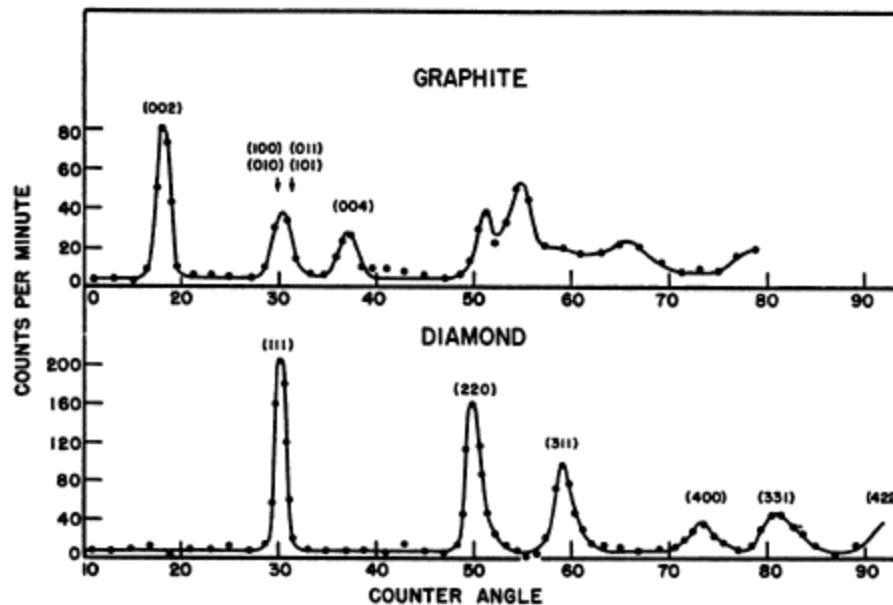
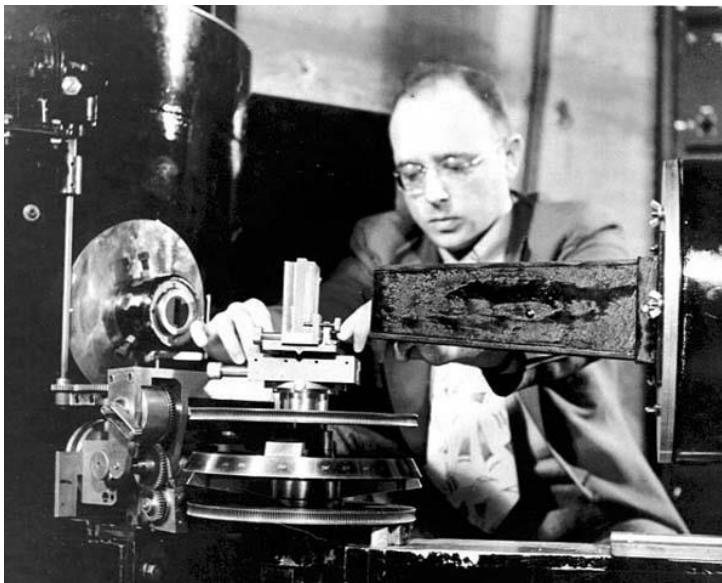


The Diffraction of Neutrons by Crystalline Powders

E. O. WOLLAN AND C. G. SHULL

Oak Ridge National Laboratory, Oak Ridge, Tennessee

(Received January 5, 1948)



So...

- See scattering where reciprocal lattice point lies on Ewald Sphere
- But, whither magnetism???



1932, Néel predicts Antiferromagnetic order

Detection of Antiferromagnetism by Neutron Diffraction*

C. G. SHULL

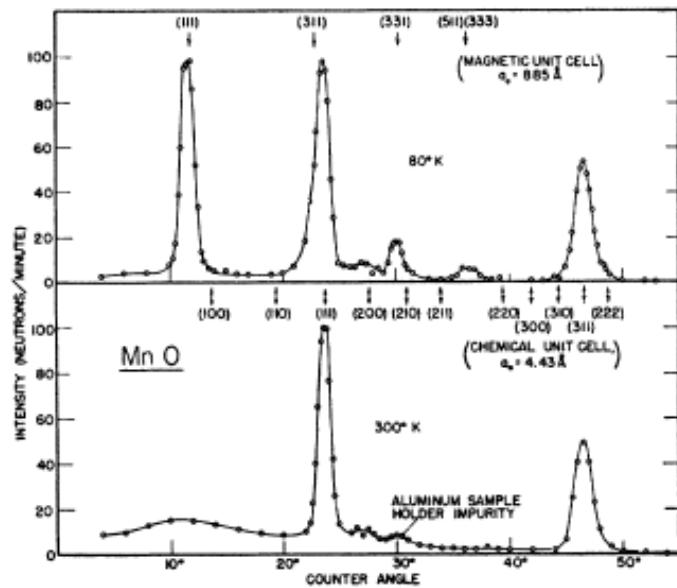
Oak Ridge National Laboratory, Oak Ridge, Tennessee

AND

J. SAMUEL SMART

Naval Ordnance Laboratory, White Oak, Silver Spring, Maryland

August 29, 1949



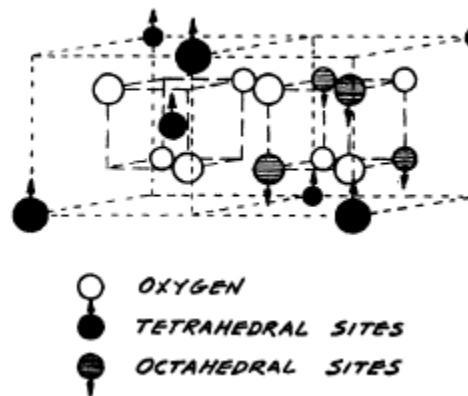
Magnetic Structure of Magnetite and Its Use in Studying the Neutron Magnetic Interaction

C. G. SHULL, E. O. WOLLAN, AND W. A. STRAUSER

Oak Ridge National Laboratory, Oak Ridge, Tennessee

December 8, 1950

Fe_3O_4 - SPINEL STRUCTURE





how**stuff**works
It's good to know



$$F(q) = \sum_i f(q) s_{+,i} e^{iq \cdot r_i}$$

$$I \propto |F|^2$$

sensitive to component of moment $\perp Q$



orthorhombic cell

OOL reflections?

HOO reflections

KOO ?

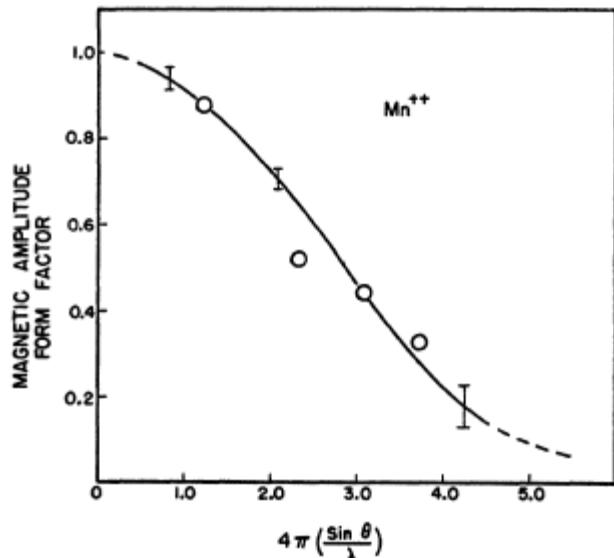


Neutron Diffraction by Paramagnetic and Antiferromagnetic Substances

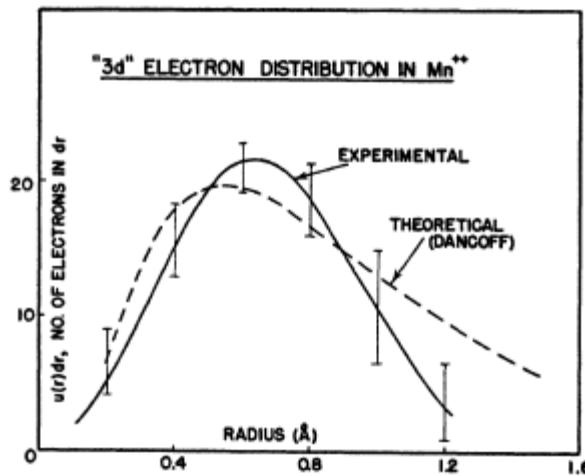
C. G. SHULL, W. A. STRAUSER, AND E. O. WOLLAN

Oak Ridge National Laboratory, Oak Ridge, Tennessee

(Received March 2, 1951)



$$u(r) = (2r/\pi) \int_0^{\infty} kf(k) \sin kr dk$$



$$d\sigma_m = \frac{2}{3} S(S+1) (e^2 \gamma / mc^2) f^2 d\Omega,$$



(a) The magnetic moments are aligned along arbitrary [100] directions, in other words, along the cube axes, as illustrated in Fig. 5. For this case it can be shown that q^2 (average) = $\frac{2}{3}$ for each of the four observed magnetic reflections in MnO.

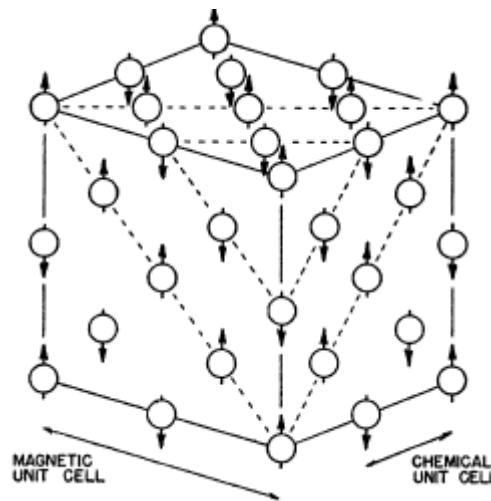


(b) The magnetic moments are aligned perpendicular to the ferromagnetic (111) sheets. For this case, q^2 vanishes for the (111) reflection and becomes 32/33 for the (311) and 32/57 for the (331) reflections.

(c) The magnetic moments are aligned arbitrarily in the ferromagnetic (111) sheets. Here q^2 should be 1 for the (111) reflection and various odd values for the other reflections, depending upon the assignment of moment orientation within the sheet.

TABLE II. Comparison between observed MnO antiferromagnetic intensities and those calculated for various models of magnetic orientation with respect to crystallographic axes.

	Calculated for various oriented models			Observed (neutrons/min)
	(a)	(b)	(c)	
(111)	1038	0	1560	1072
(311)	460	675	...	308
(331)	129	109	...	132
(511) {	54	24	...	70
(333) }				

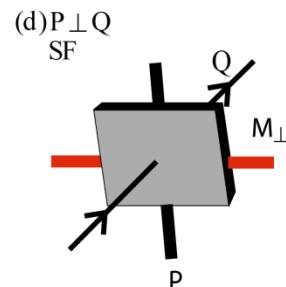
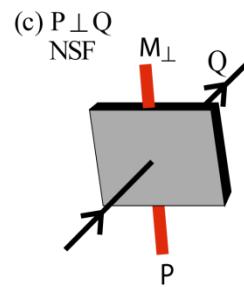
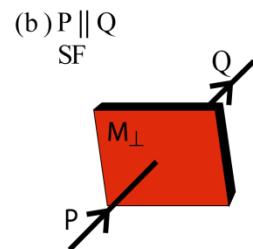
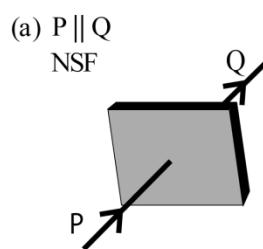


Neutron Scattering and Polarization by Ferromagnetic Materials

C. G. SHULL, E. O. WOLLAN, AND W. C. KOEHLER

Oak Ridge National Laboratory, Oak Ridge, Tennessee

(Received August 20, 1951)



Neutron Diffraction Study of the Magnetic Properties of the Series of Perovskite-Type Compounds $[(1-x)\text{La}, x\text{Ca}]\text{MnO}_3$ [†]

E. O. WOLLAN AND W. C. KOEHLER
Oak Ridge National Laboratory, Oak Ridge, Tennessee
(Received May 9, 1955)

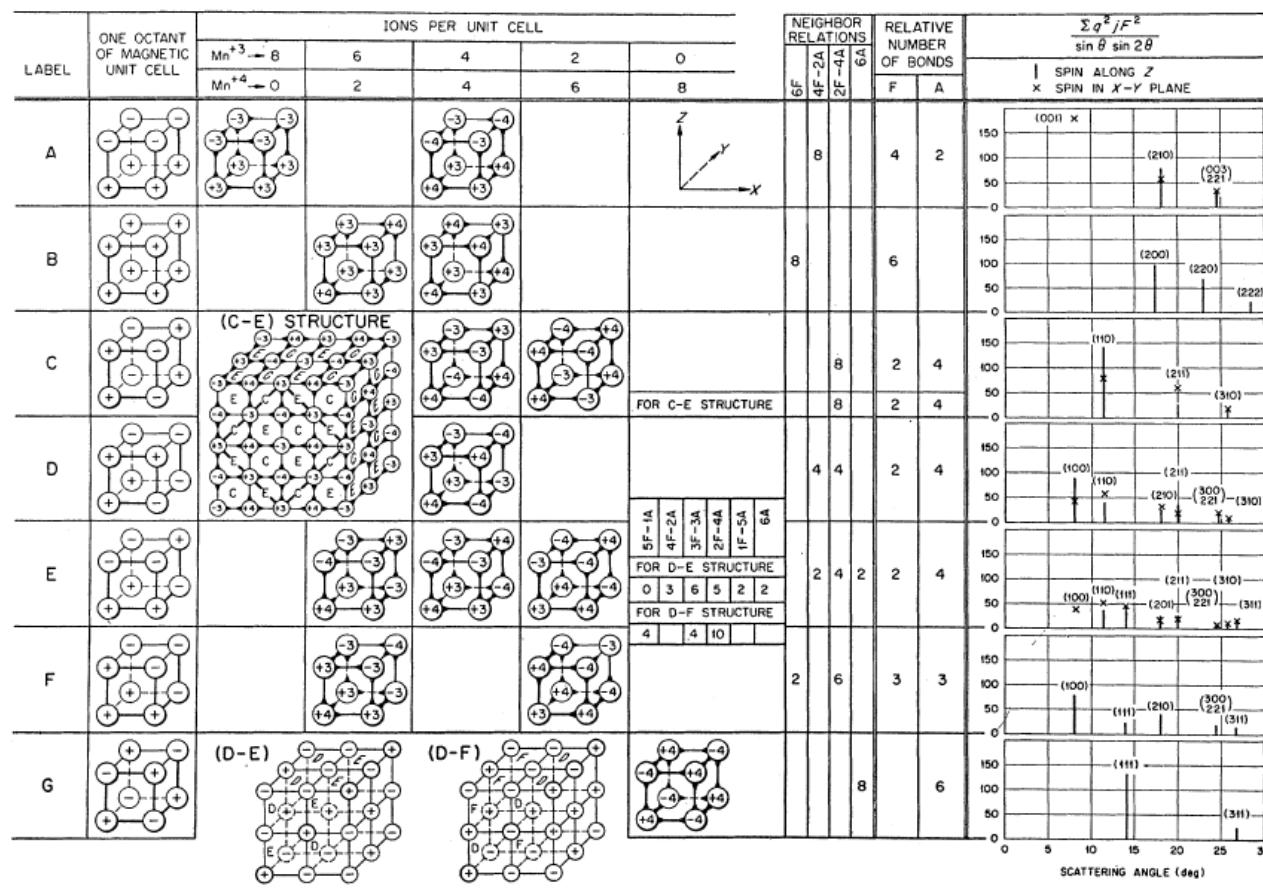


FIG. 18. Scheme of magnetic structures and related information. A, B, C, G, and (C-E) definitely observed and some evidence for D and F. Ion ordering schemes represent arrangements consistent with certain coupling criteria. Arrowheads are a schematic representation of Goodenough's semicovalent exchange coupling.



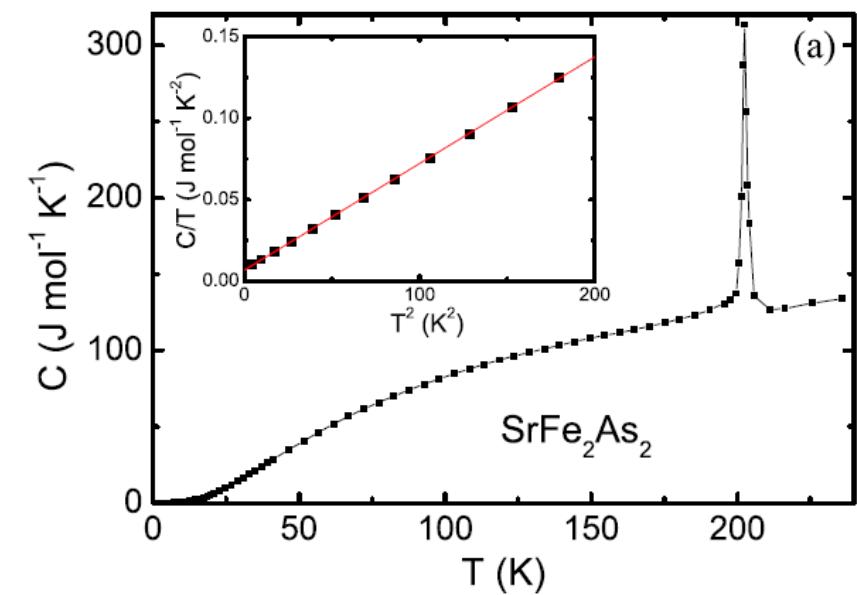
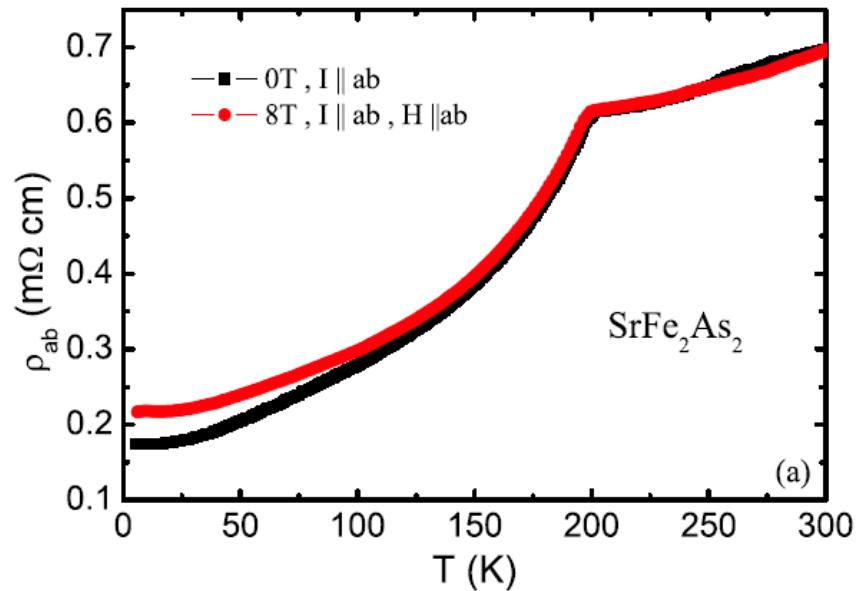
Intermission



How it starts?



Bulk Data Comes In

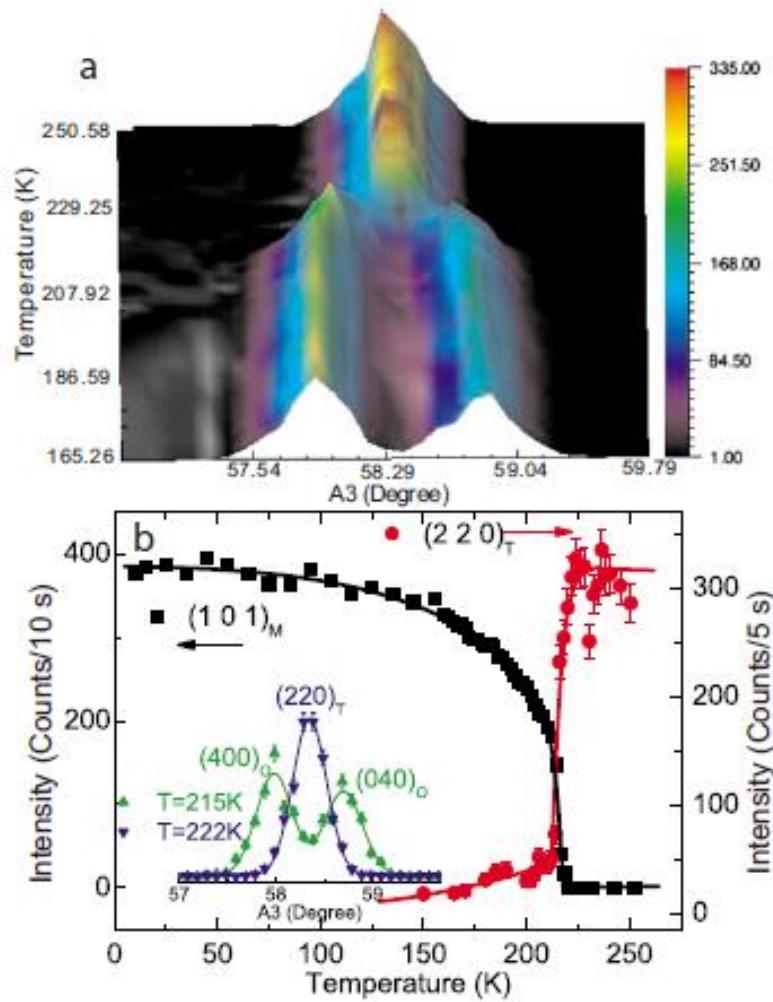


Physical Review B **78**, 22514 (2008)





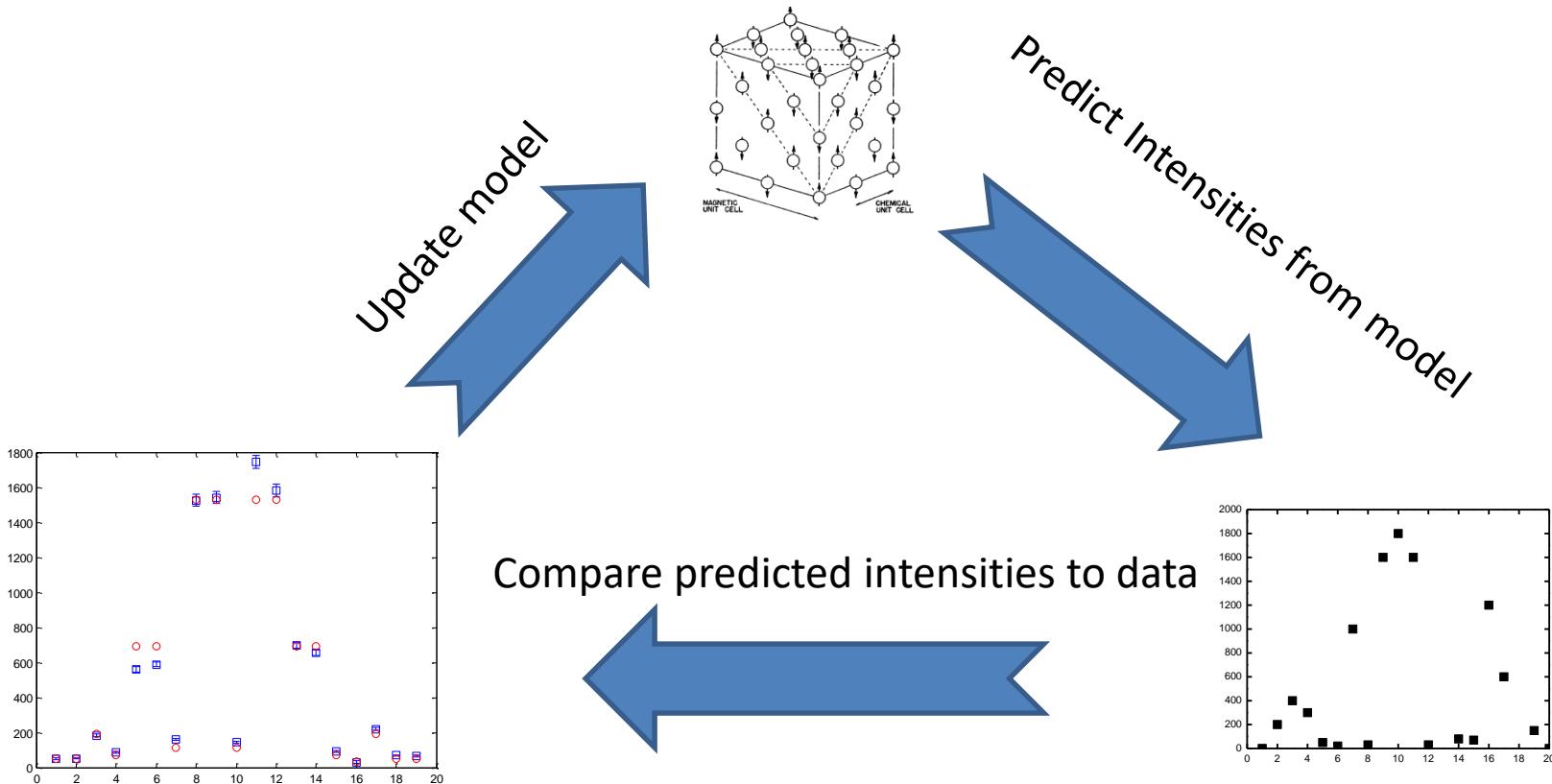
Neutrons to the Rescue



Physical Review B 78, 140504 (2008)



Guess and Check (Refinement)



Choices



Goodness of Fit

$$\chi^2 = \sum_1^n w_i (y_i - y_{ci}(\alpha))^2 \quad w_i = \frac{1}{\sigma_i^2}$$

or

$$R_{wp} = 100 \left(\frac{\sum_1^n w_i (y_i - y_{ci}(\alpha))^2}{\sum_1^n w_i (y_i)^2} \right)^{\frac{1}{2}}$$

Optimization Approaches

Simulated Annealing

Marquardt-Levenberg

Genetic Algorithm



How I proceed

- Think about the problem
- Powder diffraction
- Think some more
- Try Representational Analysis (or Group theory)
- Single crystal diffraction
- Think a lot!!!
- Polarized diffraction
- Spherical polarimetry
- Think some more...



YMn₂O₅

PRL 96, 097601 (2006)

PHYSICAL REVIEW LETTERS

week ending
10 MARCH 2006

Ferroelectricity Induced by Acentric Spin-Density Waves in YMn₂O₅

L. C. Chapon,¹ P. G. Radaelli,^{1,2} G. R. Blake,^{1,3} S. Park,⁴ and S.-W. Cheong⁴

Powder

Journal of the Physical Society of Japan
Vol. 76, No. 7, July, 2007, 074706
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Spiral Spin Structure in the Commensurate Magnetic Phase of Multiferroic RMn₂O₅

Hiroyuki KIMURA*, Satoru KOBAYASHI¹, Yoshikazu FUKUDA, Toshihiro OSAWA,
Youichi KAMADA, Yukio NODA, Isao KAGOMIYA², and Kay KOHN³

xtal

PHYSICAL REVIEW B 78, 245115 (2008)

Spiral spin structures and origin of the magnetoelectric coupling in YMn₂O₅

J.-H. Kim,¹ S.-H. Lee,^{1,*} S. I. Park,² M. Kenzelmann,³ A. B. Harris,⁴ J. Schefer,³ J.-H. Chung,⁵ C. F. Majkrzak,⁶ M. Takeda,⁷ S. Wakimoto,⁷ S. Y. Park,⁸ S.-W. Cheong,⁸ M. Matsuda,⁷ H. Kimura,⁹ Y. Noda,⁹ and K. Kakurai⁷

Xtal+spherical polarimetry

PHYSICAL REVIEW B 79, 020404(R) (2009)

Incommensurate magnetic structure of YMn₂O₅: A stringent test of the multiferroic mechanism

P. G. Radaelli,^{1,2} C. Vecchini,^{1,3} L. C. Chapon,¹ P. J. Brown,⁴ S. Park,⁵ and S.-W. Cheong⁵

Xtal+more representation analysis



Powder Diffraction

Advantages

- You get the big picture
- Can get the propagation vector
- Avoids the muss and fuss of extinction
- It's often Good Enough™

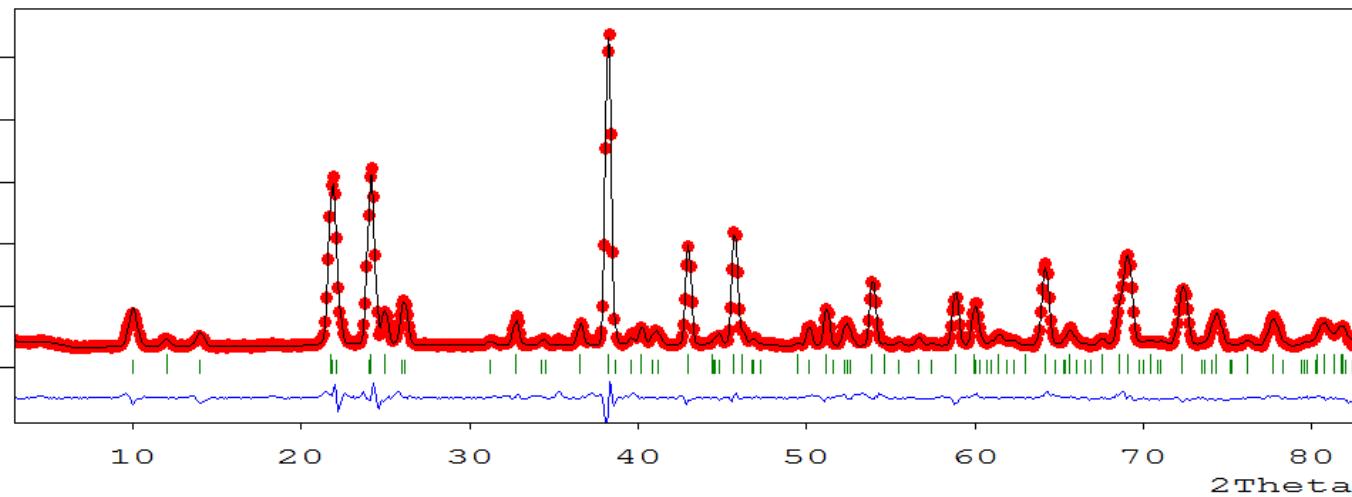
Disadvantages

- Can be hard to truly index \mathbf{k} —is it $[3\ 4\ 0]$ or $[0\ 0\ 5]$?
- You average over all symmetry equivalent \mathbf{k} at any particular Bragg angle
- You lose information in the powder averaging
- No domain info
- No multi- \mathbf{k} info
- Can be very hard to determine phase



Cycle: 2 Chi2: 18.2 dy.dat

Intensity (arb. units)



Single Crystal Diffraction

Advantages

- Can fully determine \mathbf{k}
- Can investigate domain populations
- Can apply probes (magnetic field, E-field, pressure, etc.) along a particular direction to see effect on magnetic ordering

Disadvantages

- Extinction
- Absorption depends on shape
- Reciprocal space is large...
- Crystal growth is hard...



Questions?

